# CHEMICAL DIFFERENCES BETWEEN ALPINE FIRS OF BRITISH COLUMBIA

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**Key Word Index**—Abies lasiocarpa, A. bifolia, A. balsamea, Pinaceae juvabione; epijuvabione, 1'E and 1'Z-dehydrojuvabione, 4'-dehydrojuvabione; 4'-dehydrojuvabione; juvabiol, epijuvabiol, 1'E-dehydrojuvabiol, chemotaxonomy.

Abstract—Alpine fir trees from Vancouver Island contained the sesquiterpenoids  $Z-\alpha$ -atlantone,  $E-\alpha$ -atlantone, lasiocarpenone (4R,1'S), lasiocarpenonol (4R,5R,1'S), juvabione (4R,1'R), epijuvabione (4R,1'S), 1'Z-dehydrojuvabione (4R), 1'E-dehydrojuvabione (4R,1'R,3'S), epijuvabiol (4R,1'S,3'S), 1'E-dehydrojuvabiol (4R,3'S), 4'-dehydrojuvabione (4R,1'R), and 4'-dehydroepijuvabione (4R,1'S). The presence of epijuvabione, 4'-dehydrojuvabione, and 4'-dehydroepijuvabione in coastal alpine fir provides a strong qualitative chemical basis for differentiating coastal alpine fir (Abies lasiocarpa) from Rocky Mountain alpine fir of central British Columbia (A bifolia).

#### INTRODUCTION

Alpine fir ranges from the Pacific coast east to central Alberta, and from the Yukon south to New Mexico. Quantitative differences in the amounts of terpenes have formed a basis for differentiating among alpine firs (Abies spp.) [1-6] and between alpine and balsam fir [A. balsamea (L.) Mill.] of Canada's boreal forest. More recently, Hunt and von Rudloff [2, 3] provided evidence for the separation of Western populations of alpine fir into coastal [A. lasiocarpa (Hook.) Nutt.] and Rocky Mountain (A. bifolia A Murr) species, with extensive introgression between them The coastal populations on Vancouver Island, British Columbia, are morphologically identical to the A. lasiocarpa, collected by David Douglas in coastal Oregon and described by Hooker, according to Hunt and von Rudloff [2, 3]. This position is supported by the recent work of Cope [4].

While this recent finding of Hunt and von Rudloff [2, 3] and Cope [4] is not universally accepted [7], we agree with and accept the separation of these alpine firs growing in British Columbia. Thus the differences reported between the coastal and the Rocky Mountain populations of alpine fir [2-4], earlier reports by Fraser and Swan [8-11], Manville and Kriz [12], and Leach and Thakore [13] on alpine fir extractives of A. lasiocarpa, actually describe work on A. bifolia.

Fraser and Swan [11] showed that epimeric 4(R)-[1,5'-dimethyl-1'(R/S)-hydroxy-2'-hexenyl]-1-cyclohexene-1-carboxylic-acid methyl esters (13) isolated from Rocky Mountain alpine fir were artifacts of their isolation procedure. These artifacts had been previously reported by Manville and Kriz [12] and Leach and Thakore [13] as extractives of alpine fir wood obtained from interior British Columbia trees and pulp mill effluent, respectively. None of these studies reported the isolation of the postulated precursor dehydroalcohol [12].

Juvabione-type compounds in the wood of Abies spp. appear to have chemotaxonomic significance. Several

analogues of these insect juvenile hormones have been isolated from European Silver fir (A alba Mill.) [14], balsam fir [15–17], Rocky Mountain alpine fir [12], and coastal alpine fir [18] Many sesquiterpenoids isolated from North American grown Abies spp. are listed in Table 1 which includes those compounds reported in this study

The separation of balsam fir from the alpine firs is relatively straightforward on the basis of geometric or positional isomers. The double bond of the side chain in diene sesquiterpenoids from A balsamea has been shown to be distal (4'-5'), and in A. bifolia, proximal (1'-2'). While geometric or positional isomers can be used to differentiate some species, the stereochemical compositions of co-isolates can provide key chemotaxonomic information. Abies balsamea and A. alba contain sesquiterpene alcohols that have either the R or S configuration at C-3' (9 and 10) [17, 14], while A. bifolia has been shown to contain only alcohols with the S configuration at C-3' [12] Abies balsamea, A bifolia and A. alba contain juvabione (5) with R chirality at C-1', they do not contain any epijuvabione (6).

We report here that structural differences of specific sesquiterpenoids found in A bifolia and A. lasiocarpa can differentiate these two true firs.

### RESULTS AND DISCUSSION

Sesquiterpenoid components and amounts isolated from each of three coastal alpine firs are listed in Table 2. Numerical assignments are in accordance with their GC elution order on DB-1. Many of these compounds have been previously isolated from alpine fir and other species [8-12, 14-21] The structural assignments were based on chromatographic behaviour, on the analysis of the mass, IR, <sup>1</sup>H, and especially <sup>13</sup>C NMR spectra of each component (Table 3), and by comparison with previously described compounds [8-12, 14-26].

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Abies Configurations balsamea\* bifolia† lasiocarpa‡ Compound (4R)Z-α-Atlantone (1) (4R) $E-\alpha$ -Atlantone (2) (4R, 1'S)Lasiocarpenone (3) Lasiocarpenonol (4) (4R, 5R, 1'S)(4R, 1'R)Juvabione (5) Epijuvabione (6) (4R, 1'S)1'Z-Dehydrojuvabione (7) (4R)1'E-Dehydrojuvabione (8) (4R)(4R,1'R,3'S)Juvabiol (9) (4R, 1'R, 3'R)Isojuvabiol (10) (4R, 1'S, 3'S)Epijuvabiol (11) (4R, 3'S)1'E-Dehydrojuvabi-3'-ol (12) (4R, 1'R/S)2'-Dehydrojuvabi-1'-ols (13)§ 4'-Dchydrojuvabione (14) (4R, 1'R)4'-Dehydroepijuvabione (15) (4R, 1'S)

(4R, 1'R, 3'S)

(4R.1'R)

Table 1. Juvabione-type compounds isolated from North American Abies spp

4'-Dehydrojuvabi-3'-ol (16)

3'-Dehydrojuvabi-5'-ol (17)§

This paper reports the first isolation of a 1'E-dehydrojuvabiol (12), the precursor which has been postulated [12] for the isolated tertiary alcohols (13) Compound 12 was obtained from the whole wood of each of three coastal alpine fir trees growing on Vancouver Island. This compound was isolated in a manner similar to that used previously in isolating the 1'-(13) [11, 12] and 5'-(17) [17] dehydro-tertiary alcohols, except that no acid-washed glassware was used and only HPLC grade solvents were used Dehydro-alcohol 12 proved to be most stable and showed no tendency to rearrange to the tertiary alcohol mixture, 4(R)-[1',5'-dimethyl-1'(R/S)-hydroxy-2'hexenyl]-1-cyclohexene-1-carboxylic acid methyl esters (13) [11, 12] No evidence of the epimeric E- or Z-dehydroalcohols was noted The basic structure of 12 and its acetate (12Ac) was determined by extensive <sup>1</sup>H and <sup>13</sup>C NMR studies (Table 3). It was confirmed by reducing a sample of 1'E-dehydrojuvabione (8) with sodium borohydride to a mixture of two (1.1) 1'E-dehydro-alcohols (12 and 12b), one of which was identical to the isolated dehydro-alcohol (12) as evidenced by their 13C NMR spectra (Table 3) This isolation provides strong evidence that the tertiary alcohols (13 and 17) (Table 1) found in the isolates from A bifolia [11, 12] and A balsamea [17] are artifacts of the isolation/purification procedures

The occurrence of only one dehydro-alcohol in the wood of *A. lasiocarpa* and the absence of any asymmetric centre at C-1' precludes the possibility of assigning the stereo-configuration at C-3' by <sup>13</sup>C NMR However, the isolation of juvabiol (9) and epijuvabiol (11) from the same wood, each with the *S* configuration at C-3', indicates (Fig.1) that for this isolated precursor dehydroalcohol (12), the configuration about C-3' is *S* We therefore

assign the structure of **12** as 4(R)-[1',5'-dimethyl-3'(S)-hydroxy-2'-hexenyl]-1-cyclohexene-1-carboxylic acid methyl ester

Rocky Mountain alpine fir was known to contain varying quantities of epimeric juvabiofs—juvabiof (9) and epijuvabiof (11) in a 10 1 ratio [17]. Coastal alpine fir (Table 1) contains minor amounts of the same two alcohols with the amount of 9 varying from ca 7% to less than 2% of the amount of 11. Each alcohol has the same S configuration at C-3′, but differs in stereochemistry at C-1′. The absence of alcohols with the R configuration at C-3′ confirms the species differentiation from A balsamea.

The S configuration at C-3' in the alcohols (9 and 11, Table 1) supports the postulated biosynthetic pathway (Fig 1) and is in general agreement with the biosynthetic pathway previously presented by Manville and Kriz [12] for Rocky Mountain alpine fir The notable difference in the current biosynthetic pathway is the presence of significant quantities of a naturally occurring epimer of juvabione (5), namely epijuvabione (6) and the presence of dehydrojuvabione (14) and 4'-dehydroepijuvabione (15).

The isolation of 7 in admixture with 5 and 6 afforded the opportunity to assign the  $^{13}$ C NMR chemical shifts of all carbons (Table 3) Comparison of the shifts of C-3, C-4, C-5 and C-6 of the E and Z pair of 1'-dehydrojuvabiones (7 and 8) and the corresponding pair of  $\alpha$ -atlantones (1 and 2) [18] indicates that the preferred conformation for 7 is similar to that of 1, as shown in Fig. 1

Epijuvabione (6) and 4'-dehydroepijuvabione (15) were not previously known to occur in the North Americangrown firs [12, 15, 16] Cerny et al [21], reported the occurrence of 6 and 15 in balsam fir, but the work was conducted on a putative 4 balsamea cross growing in a

<sup>\*</sup>A balsamea (L) Mill, balsam fir

<sup>†</sup>A bifolia A Murr, Rocky Mountan alpine fir

<sup>‡</sup>A lasiocarpa (Hook) Nutt, coastal alpine fir

<sup>√</sup> Compound present in the wood. nCompound not looked for

<sup>§</sup>These compounds are artifacts, the preceding compound is the extractive present in the wood

<sup>-</sup> Not present in detectable amounts

Table 2. Abies lasiocarpa, lipophilic extractives.

	Site*	Age			% composition (GC) of sesquiterpenoid hexane solubles												
Tree			Yield (% o.d. wood)		Un†	1	2	3	4	5,6,7 R <sub>t</sub> (min	8	9,11	12	14,15			
			C <sub>6</sub> H <sub>6</sub> /EtOH	Hexane	7.3	7.4	79	8.3	8.8	9.8	10.0	10.1	10.2	10.3			
 1	1	73	2.52	0.72	4.0	43	35.7	8.0	4.0	11.5	6.2	3 1	8.1	0.8			
2	2	36	2.60	0.78	27	4.8	32.9	3.6	2.0	8.3	19.7	58	10.6	0.9			
3	2	56	2.38	0 93	1.5	2.4	159	29 5	10.2	6.3	6.9	3.1	8 2	5.7			

<sup>\*</sup>Site locations: (1) Mount Washington, west of Courtenay, B.C., (2) Green Mountain, southwest of Nanaimo, B.C. †Un, unknown compound.

Table 3 <sup>13</sup>C NMR chemical shifts (δ) in CDCl<sub>3</sub>

	C*														
Compound†	1	2	3	4	5	6	7	1'	2'	3′	4′	5′	6′	7′	8′
Juvabione (5)	129 9	138.8	29.4	37.5	24 5	24 5	167.2	32.4	47 4	209 7	52 2	24 2	22.3	22.3	16.2
Epijuvabione (6)	129 9	1388	28.2	37.4	25.8	246	167.2	32.6	47.5	209.7	52.1	24.2	22 2	22.2	16.1
Z-1'-Dehydrojuvabione (7)	130 1	138.7	29 6	35.1	26.4	249	167.5	160.2	124.9	201 2	53.3	24.3	22.4	22.4	20.2
E-1'-Dehydrojuvabione (8)	1300	1380	30 4	43 2	26.5	25.0	167 4	159.9	122.6	201.2	53.4	24 2	22.4	22.4	17.2
E-1'-Dehydrojuvabiol (12)	130 1	139.0	31.0	418	31.0	29 7	1678	141 1	1279	66.9	470	24 7	22.6	24.7	147
E-1'-Dehydroisoiuvabiol (12b)	130.1	1390	31.0	417	310	29.7	1678	141 0	1278	66.8	47.0	246	230	246	14.6
E-1'-Dehydrojuvabiol acetate															
(12Ac)‡	1299	1388	308	41 5	26.8	24.4	1677	142.7	123.4	700	44.0	24.5	22 7	22.5	14.8
4'-Dehydrojuvabione ((14)	130.0	139.1	29 5	37 5	24.6	24.6	167.7	33.1	48.5	200.6	123 8	1550	20.4	27.4	16.3
4'-Dehydroepijuvabione(15)	130.0	139 1	28.2	37 5	259	24.7	1677	33 3	48.7	200.6	123.8	1550	20.4	27 4	16 1

<sup>\*</sup>See 1-17 for position numbering.

<sup>†</sup>The methyl ester carbon resonates at  $\delta 51.3 \pm 0.2$  ppm. ‡The acetate carbons are found at  $\delta 169.9$  and  $21.0 \pm 0.2$  ppm

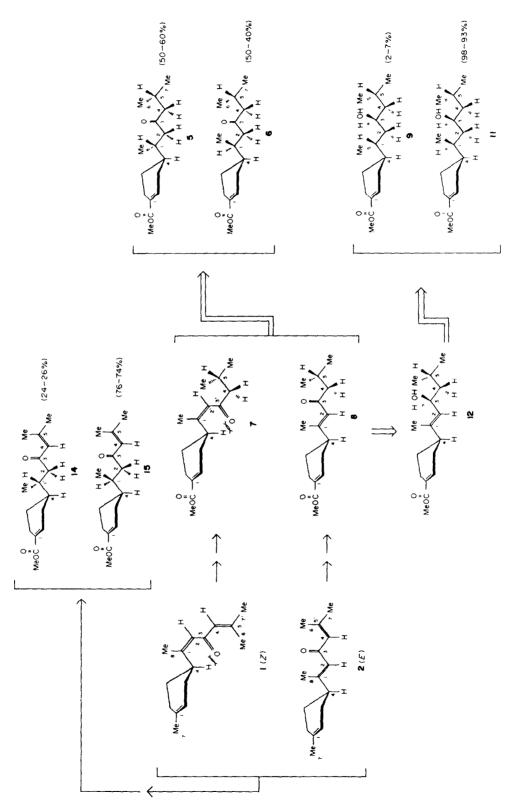


Fig 1 Postulated biosynthetic pathways for juvabione-type sesquiterpenoids of coastal alpine fit.

Me 
$$\frac{1}{1}$$
  $\frac{1}{1}$   $\frac$ 

European arboretum [16, 19]. The presence of 5 in A. bifolia and 5, 6, 14 and 15 in the whole wood of Vancouver Island grown alpine firs provides the basis for a chemotaxonomic differentiation between A. lasiocarpa and A. bifolia.

## **EXPERIMENTAL**

Isolation. The compounds used for this study were isolated from air-dried, debarked whole wood of branches from coastal alpine firs from Green Mountain and Mount Washington on Vancouver Island using methods described elsewhere [9, 12, 16, 17].

General. The <sup>1</sup>H and <sup>13</sup>C NMR spectra were determined on samples dissolved in CDCl<sub>3</sub> ( $\delta^1 H = 7.26$  at 400 MHz and  $\delta^{13} C$ =76.88 at 100.6 MHz) at ambient temperatures (21°) using a Brucker 400 MHz NMR spectrometer. Samples for proton spectra were about a factor 20 more dilute than those for the carbon spectra Low resolution mass spectra were determined on a Finnigan GC/MS at the Institute of Ocean Sciences, Pat Bay and on a Finnigan mass spectrometric detector (ION-TRAP<sup>R</sup>). GC analyses were performed on a Spectra-Physics model 7100 GC equipped with a flame ionization detector (FID) and the ION-TRAP. The columns used were 30 m  $\times$  0.25 mm (i.d.) DB-1 fused silica columns The column oven was programmed as follows: 200° (1 min), 10° per min to 275° (15 min). The first split/splitless injector was maintained at 250°, the second was a J&W cool on-column injector, for the ION-TRAP detector. The FID detector was kept at 295°. The carrier gas was maintained at 150 psig in the split/splitless injector port and at 11 psig in the cool on-column injection port. Butane gas had a linear velocity of 22 cm<sup>-1</sup>. Extractions and separations were similar to those reported earlier [9, 12, 16, 17], except for the purification of 4 which was accomplished by Al<sub>2</sub>O<sub>3</sub> CC using 1% EtOH in benzene [see 18]

Z-α-Atlantone (1). Identification of 1 was made on the basis of its <sup>1</sup>H, <sup>13</sup>C NMR, IR and MS comparison with published information [12, 18, 22-25]

E-α-Atlantone (2). Identification of 2 was made on the basis of its <sup>1</sup>H, <sup>13</sup>C NMR, IR and MS and comparison with published information [12, 18, 22–25].

Lasiocarpenone (3). Identification of 3 was made on the basis of its <sup>1</sup>H, <sup>13</sup>C NMR, IR and MS and comparison with published information [10, 18].

Lasiocarpenonol (4). See ref. [18] for the details of the identification of 4 which was made on the basis of its <sup>1</sup>H, <sup>13</sup>C NMR, IR and MS and companson with 3.

Juvabione (5), epijuvabione (6) and 1'Z-dehydrojuvabione (7) Identifications of 5-7 were made on the basis of their <sup>1</sup>H, <sup>13</sup>C NMR (Table 3), and comparison with published information [12-21].

1'E-Dehydrojuvabione (8). Identification of 8 was made on the basis of its  $^{1}$ H and  $^{13}$ C NMR, IR and MS and published information and comparison with other structurally related compounds [12-21]  $^{1}$ H NMR:  $\delta$ 6.94 (1H, m, H-2), 6 01 (1H, brs, H-2'), 3 68 (3H, s, -OMe), 2.44 (1H, brd, J = 17.5 Hz,H-3e), 2.25 (2H, d, J = 7 1 Hz, H-4'), 2.07 (3H, d, J = 1 Hz, C-1' -Me), 2.05 (1H, m, H-5'), 1.83 (1H, m, H-5e), 1.49 (1H, dtd, J = 5 5, 11 and 13 Hz, H-5a), 0.87 [6H, C-5'-(Me)<sub>2</sub>], 2.3-1.9 (6H);  $^{13}$ C NMR: see Table 3; MS. no  $[M]^+$ , 232 (3), 207 (80), 204 (31), 175 (42), 147 (100), 137 (21), 119 (60), 105 (30), 95 (75), 91 (31), 85 (60), 79 (21), 77 (27), 69 (95), 67 (35), 57 (32) and 41 (30).

Juvabiol (9) and epijuvabiol (11). Identifications of 9 and 11 and their acetates were made on the basis of their <sup>1</sup>H, <sup>13</sup>C NMR, and comparison with published information [12–14, 17, 26].

1'E-Dehydrojuvabi-3'-ol (12). <sup>1</sup>H NMR  $\delta$  7.00 (1H, m, H-2), 5.20 (1H, br d, J = 6.5 Hz, H-2'), 4.5 (1H, dd, J = 6 and 8 Hz, H-3'), 3 73 (3H, -OMe), 1.70 (1H, d, J = 2.1 Hz, 1'-Me), 0.93 (3H, d, J = 6.5 Hz, 5'-Me), 0.92 (3H, d, J = 6.5 Hz, 5'-Me) and 2.5-1.4

(11H), <sup>13</sup>C NMR see Table 3, MS. no [M]<sup>+</sup>, 234 (5), 209 (7), 207 (6), 177 (66), 150 (28), 149 (40), 139 (43), 134 (21), 131 (17), 127 (19), 121 (21), 107 (39), 105 (28), 97 (32), 93 (50), 91 (35), 85 (100), 81 (47), 79 (70), 77 (33), 71 (95), 69 (42), 67 (38), 59 (34), 57 (99), 55 (39), 53 (48), 43 (77), and 41 (98)

Acetylation of 12 Compound 12 was dissolved in pyridine and Ac<sub>2</sub>O and left at room temp overnight to afford 1'E-dehydrojuvabi-3'-ol acetate (12Ac)  $^{1}$ H NMR  $\delta$  6 97 (1H, m, H-2), 5 58 (1H, m, H-3'), 5 08 (1H, br d, J=90 Hz, H-2'), 3 72 (3H, s, -OMe), 2 43 (1H, br d, J=18 Hz, H-3e), 2 27 (1H, m, H-6e), 2 22 (1H, m, H-3a), 2 11 (1H, m, H-4), 2 10 (1H, m, H-6a), 2 00 (3H, s, -Ac), 1 82 (1H, m, H-5e), 1 75 (3H, d, d) = 2 1 Hz, 1' -Me), 1 55 (2H, d), H-4'e and H-5a), 1.47 (1H, d), H-5'), 1 30 (1H, d), H-4'a), 0 90 (3H, d), d) = 6 5 Hz, 5' -Me), 0 88 (3H, d), d) = 6 5 Hz, 5' -Me); d<sup>13</sup>C NMR see Table 3.

Reduction of 8 with NaBH<sub>4</sub>. The 1'E-Dehydrojuvabiols, compounds 12 and 12b, were obtained by NaBH<sub>4</sub> reduction of 8 as a 1.1 mixture  $^{-1}$ H NMR of 12 and 12b was similar to that of 12.  $^{-13}$ C NMR see Table 3, MS no [M] $^{+}$ , 234 (5), 209 (8), 207 (7), 177 (65), 150 (30), 149 (37), 139 (41), 121 (21), 107 (40), 105 (25), 97 (30), 93 (50), 91 (30), 85 (100), 81 (45), 79 (70), 69 (42), 67 (38), 59 (34), 57 (98), 55 (38), 43 (77) and 41 (95)

4'-Dehydrojuvabione (14) and 4'-dehydroepijuvabione (15). Identification of 14 and 15 were made on the basis of MS, <sup>1</sup>H, <sup>13</sup>C NMR, and comparison with published information [12-14, 16, 19-21]

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